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ASSESSMENT OF ELECTROCOAGULATION PROCESS. APPLICATION IN ARSENIC REMOVAL FROM DRINKING WATER

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Abstract

Arsenic removal from groundwater as drinking water source was investigated using electrocoagulation (EC) process with horizontal anodes of iron plate under galvanostatic regime. The current density of 5 A·m⁻² was determined as the optimum operating conditions at the electrolysis time of maximum 25 minutes and pH ranged between 6.5 to 8.5, by which the arsenic concentration was reduced below 10 μg·L⁻¹. The aspects related to the arsenic removal mechanism were discussed by Zeta potential. This iron anode-based EC process was successfully applied for the removal of arsenic from a real groundwater characterized by high arsenic concentration of 255.80 μg·L⁻¹ under aeration conditions. After 15 minutes of electrolysis the arsenic removal efficiency of 99.59% was achieved under the optimum operating conditions with the energy consumption of 0.006 kWh·m⁻³ that is 100 times lower than the reported in the literature.

Key words: arsenic removal, drinking water treatment, electrocoagulation, iron anodes

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